

# Lithium-Metal Anodes: Problems and Multiple Solutions Based on Electrolytes, Hosts, and Protective Layers

**Ji-Guang Zhang (PI), Wu Xu, Jun Liu**

*Pacific Northwest National Laboratory*

2018 DOE Vehicle Technologies Program Review  
June 18-21, 2018

This presentation does not contain any proprietary, confidential, or otherwise restricted information

# Overview

## Timeline

- Project start date: Oct. 2016
- Project end date: Sep. 2021
- Percent complete: 30%

## Budget

- Total project funding \$50M
  - DOE share 100%
- Funding received in FY17: \$10M
- Funding received in FY18: \$10M

## Barriers

- Low Coulombic efficiency
- Li dendrite growth
- Large volume change

## Partners

- Pacific Northwest National Laboratory
- Stanford University
- SLAC
- Idaho National Laboratory
- University of Washington
- University of California at San Diego
- UT Texas

# Relevance/Objectives

- Develop stable electrolytes to improve the Coulombic efficiency of Li metal anode
- Develop ex situ formed artificial SEI layer to protect Li metal against dendrite growth
- Develop efficient host for Li metal to minimize volume change of Li based anode
- Enable operation of thin Li metal in lean electrolyte conditions
- Enable high efficiency and safe utilization of Li metal anode for high energy density Li metal batteries required for long range EV applications

# Milestones

- Investigate methods to extend the cycling and stability of Li metal pouch cells.. (Dec-31, 2017). **Completed**
- Establish the high Ni NMC coin cell properties using the materials synthesized by the team and supplied by other sources (Mar-31, 2018). **Completed**
- Provide feedback on characterization of the new materials and concepts by the characterization team. (Jun-30, 2018). **On track**
- Develop and implement methods to improve and understand cycle and calendar life limitations of pouch cells. (Sep-30, 2018). **On track**

# Approach

## Battery500 Solutions

### Stable Electrolyte (PNNL)

- High concentration electrolyte



- Localized High Concentration Electrolyte (LHCE)



- Non-flammable LHCE

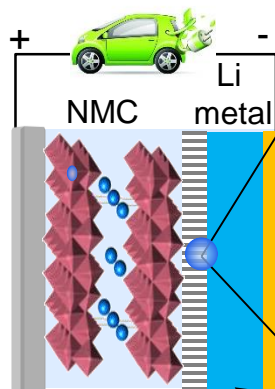


- ≥300 Wh/kg Li metal pouch cells have been successfully fabricated and demonstrated >100 stable cycling.

### Host:

- Form stable host for Li metal: Li-SiO composite (Stanford)
- Thin stable 3D conductive host without sacrificing the energy density (PNNL)

### Challenges on Li metal anode



Low CE leads to consumption of Li and electrolyte

Dendrite leads to short circuit

Large Volume change leads to mechanical and electrical failure

### Simulation

Li metal batteries (UW)

### Testing

Li metal batteries (INL)

### Protection layer:

- Self-healing polymer (Stanford)
- Form stable artificial SEI layer via gas-phase reaction (Stanford)
- Selective deposition and stable encapsulation of lithium through heterogeneous seeded growth (Stanford)
- Li-methyl Carbonate Protection Layer (UCSD)
- Solid state electrolyte (UT Austin)

### Characterization:

- Reveal the atomic structure of Li metal by cryo-electron microscopy (Stanford/UCSD)
- X-ray characterization (SLAC)

# Technical Accomplishments

## Localized high concentration electrolyte (LHCE) for high efficiency Li metal batteries

Limitations of HCE:

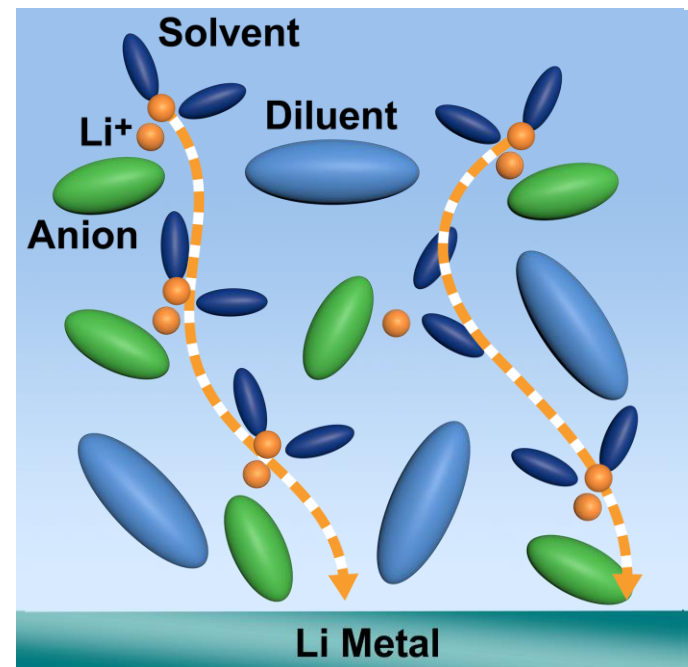
- High cost
- High viscosity
- Low conductivity

Rational design of LHCE electrolyte:

- a. A base solvent (DMC) with a high solvability of Li salt
- b. A Li salt (LiFSI) stable with Li metal anode
- c. A diluent (BTFE) with a very limited solvability of Li salt and fully mixable with base solvent
- d.  $\text{Li}^+$  will transport along the solvent/salts clusters with localized high salt concentration so advantages of HCE can be preserved.

Advantages of LHCE:

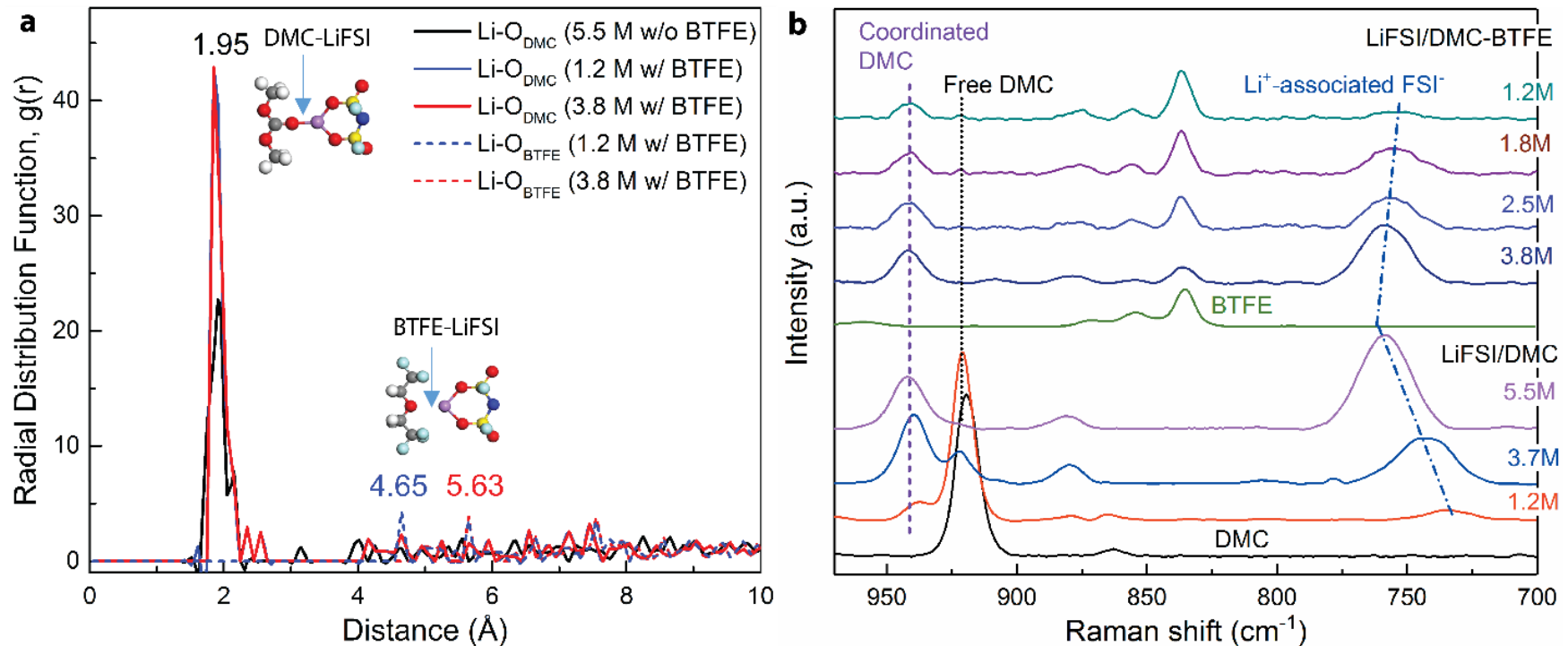
- Retained all advantages of HCE
- Low cost
- Low viscosity
- High conductivity



# Technical Accomplishments

## Formation of localized high concentration clusters in molecular level

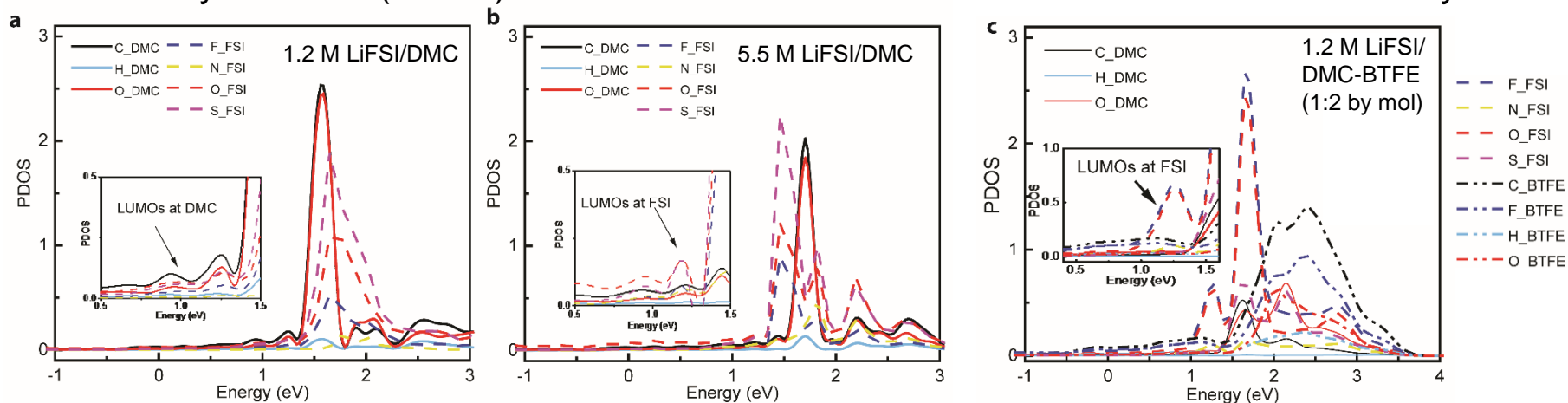
A typical example of LHCE: 1.2 M LiFSI in DMC:BTFE (1:2 by mole)



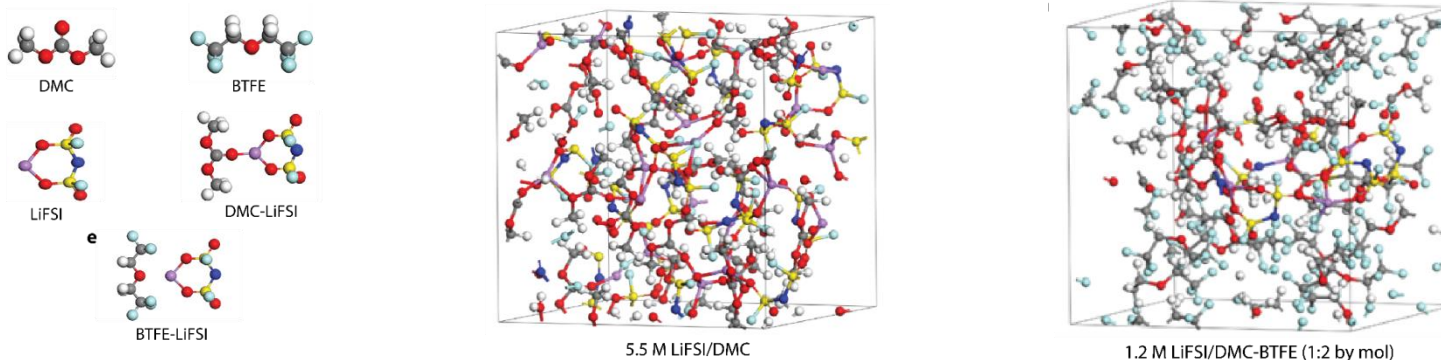
- a. DMC-LiFSI has binding that is much stronger than BTFE-LiFSI binding
- b. Addition of BTFE does not significantly affect the original LiFSI-DMC binding

# Technical Accomplishments

Projected density of states (PDOS) of each atom on the Li anode surface in different electrolytes



Snapshots of electrolyte/salt mixtures from AIMD simulations at 30 °C

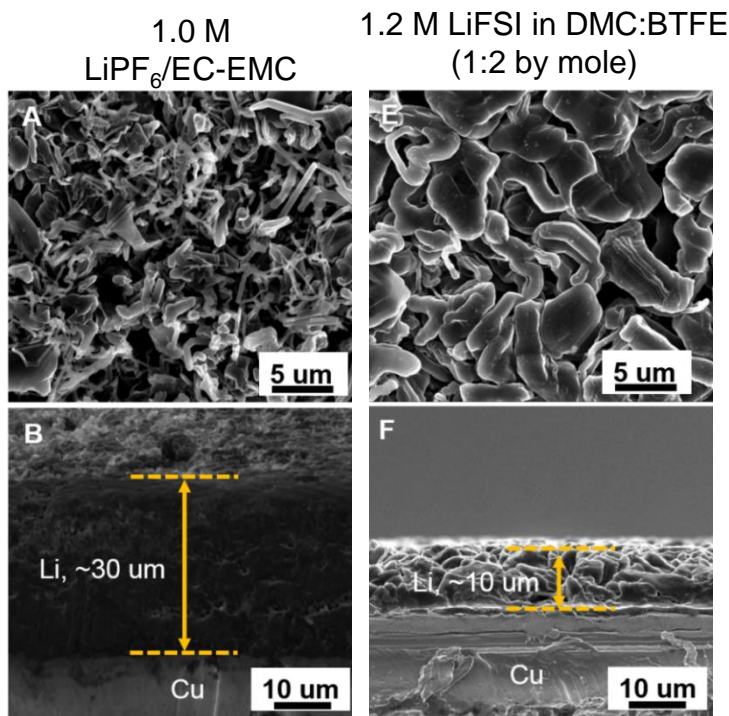


- LUMO energies of FSI anion in HCE and LHCE shift to lower than those of DMC solvent so salt will be decomposed first to form a LiF rich SEI to protect Li metal anode.
- All LiFSI salt molecules are closely coordinated with DMC instead of BTFE in the HCE and LHCEs.

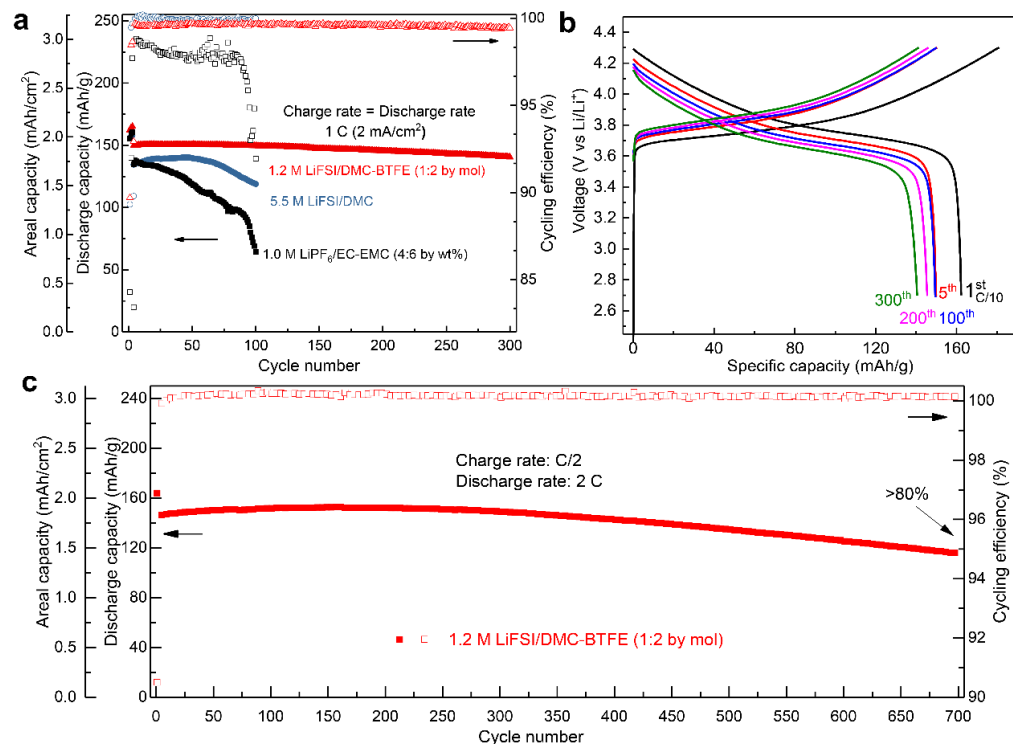


# Technical Accomplishments

## High performance Li metal batteries based on LHCE



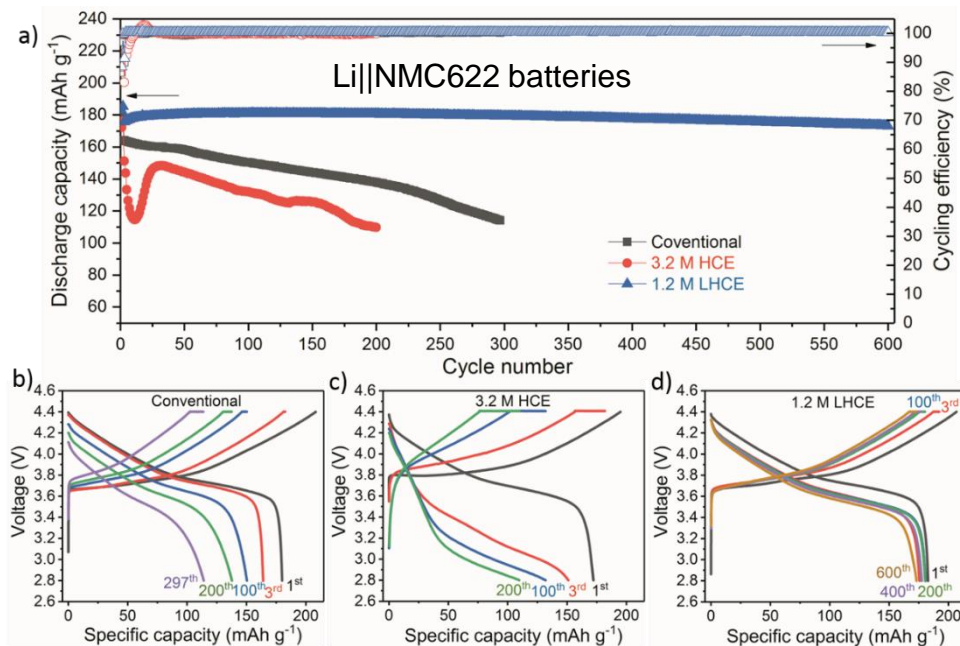
*Highly packed granule Li particles (~10  $\mu\text{m}$ ) formed in LHCE electrolyte exhibits small surface area and will not penetrate separator*



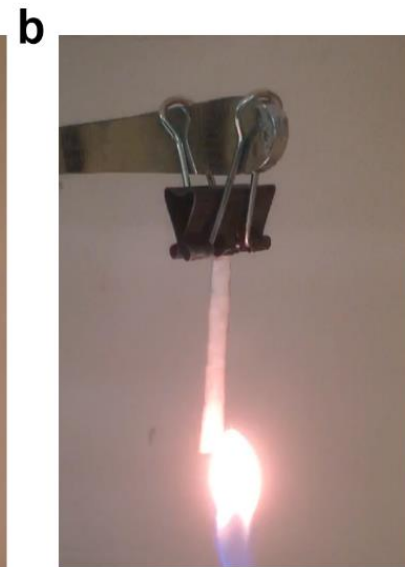
*Stable cycling of Li/NMC batteries with LHCE (>80% capacity retention after 700 cycles at a high current density of  $2.0 \text{ mA}/\text{cm}^2$ )*

# Technical Accomplishments

## Nonflammable LHCE for high-performance LMBs



a). LiPF<sub>6</sub> in carbonates



b). EL-313

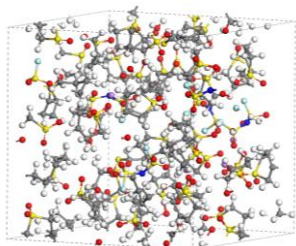
- High CE of LMAs and good cycle life of Li||NMC622 cells can be achieved with nonflammable EL-313.

# Technical Accomplishments

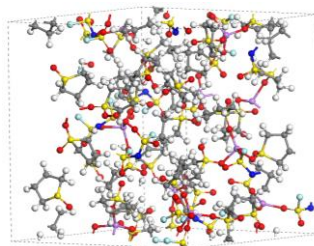
## Localized high concentration electrolyte LiFSI-3TMS diluted by TTE

a. Solvation structure in HCE and is retained with addition of non-solvating TTE.

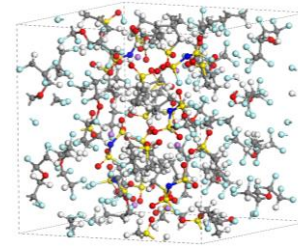
Dilute (LiFSI-8TMS)



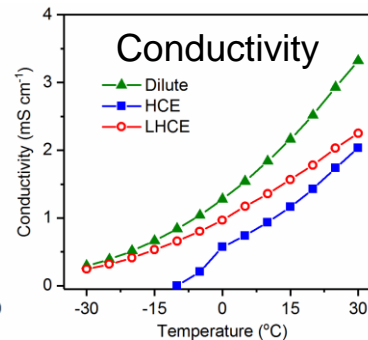
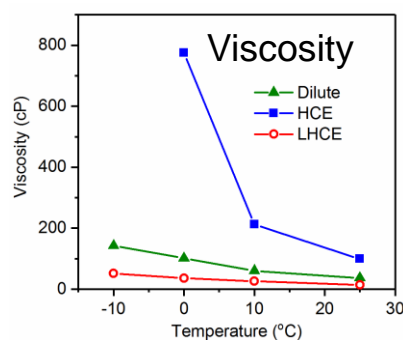
HCE (LiFSI-3TMS)



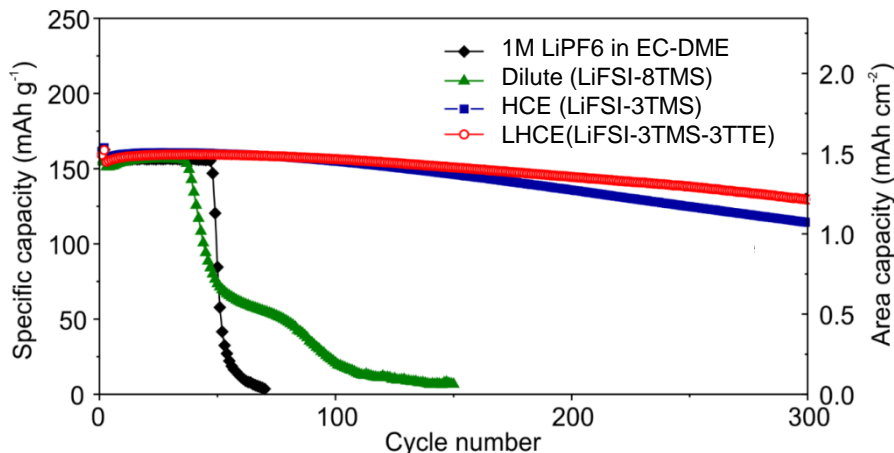
LHCE (LiFSI-3TMS-3TTE)



b. LHCE has a lower viscosity and higher conductivity.

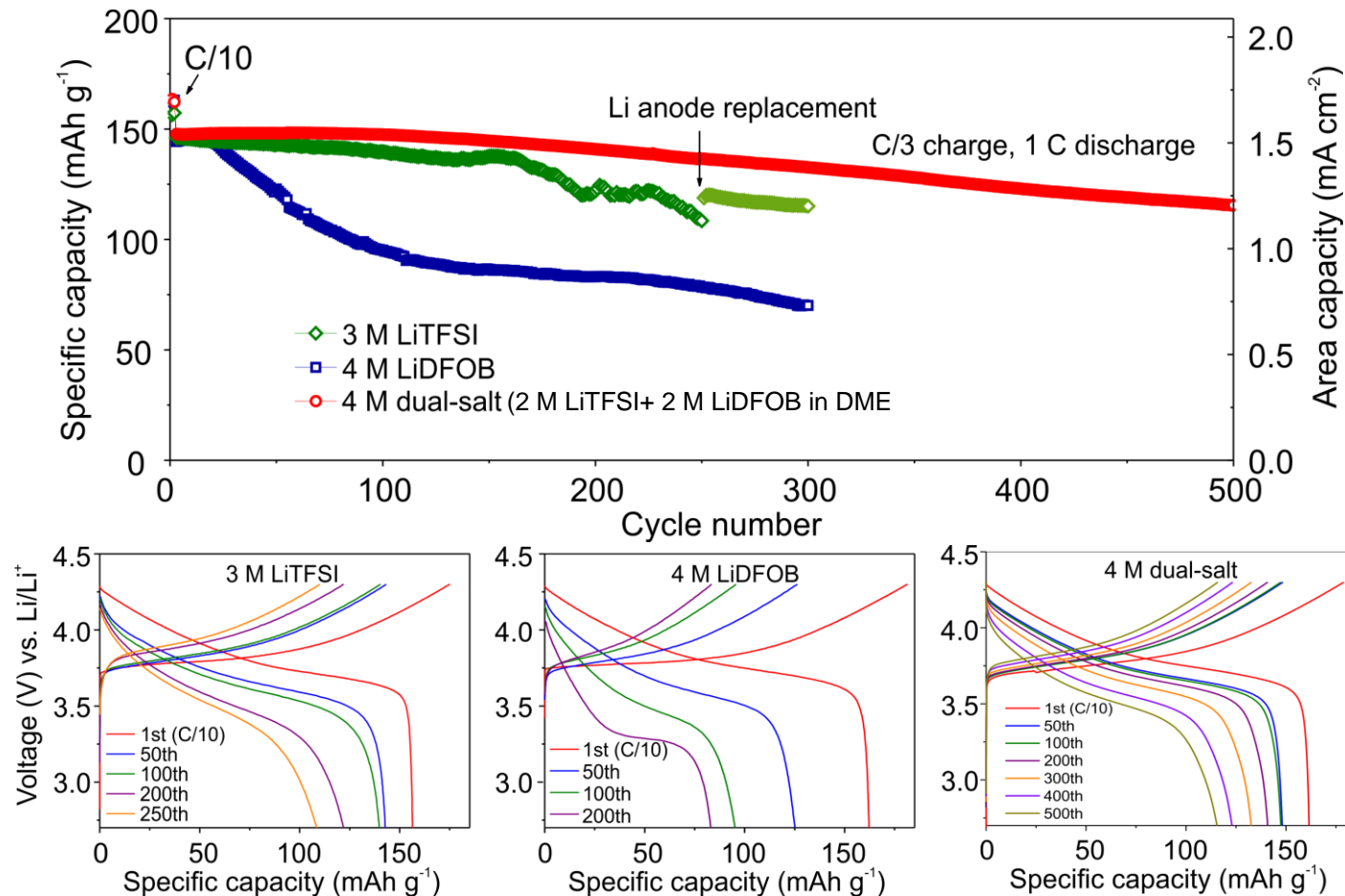


c. LHCE has a lower viscosity and higher conductivity.



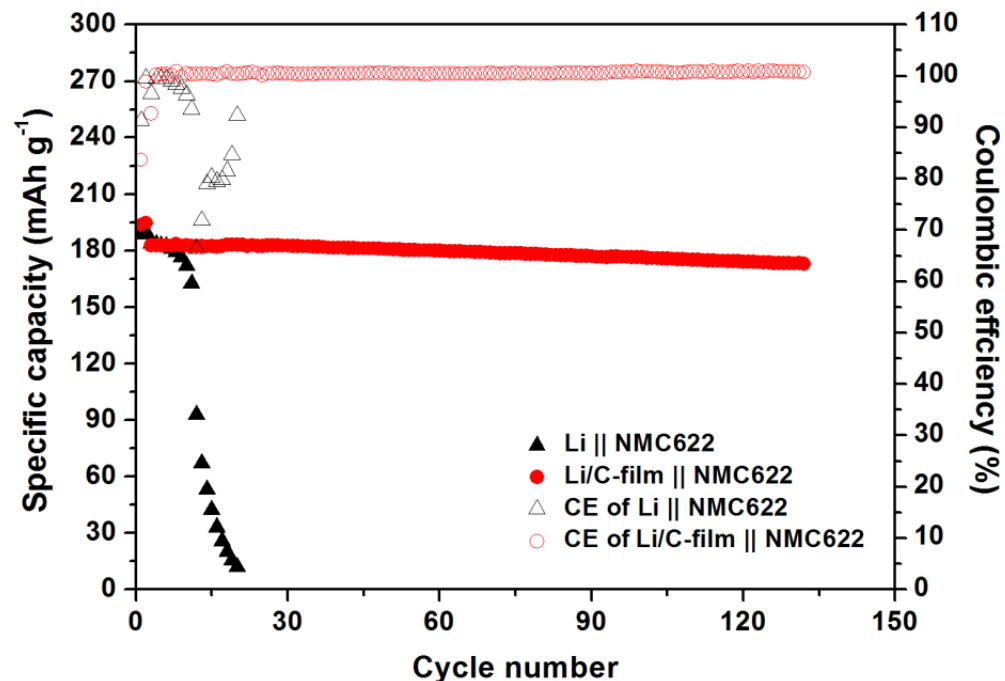
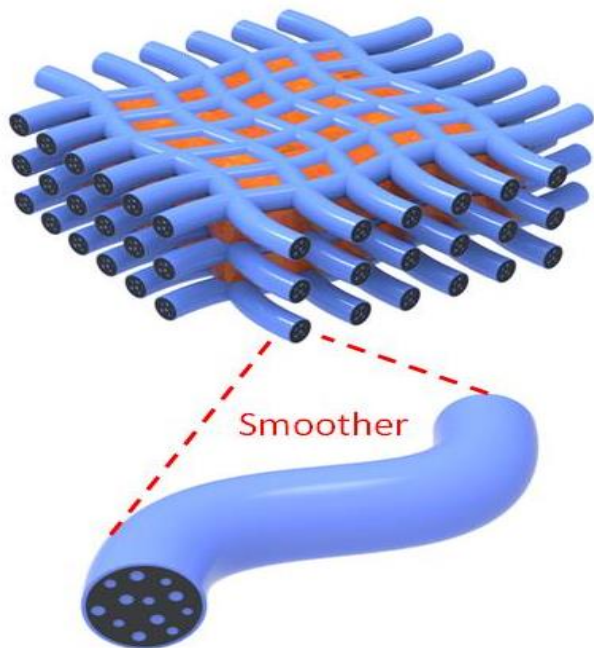
# Technical Accomplishments

## Li || NMC333 cycling performance



➤ Dual-salt ether electrolyte leads to good cycling stability of Li||NMC cells.

# Technical Accomplishments



- Thin stable 3D architectures without sacrificing the energy density
- Controlled Li amount and low porosity
- Self-healing to prevent inhomogeneous Li metal deposition

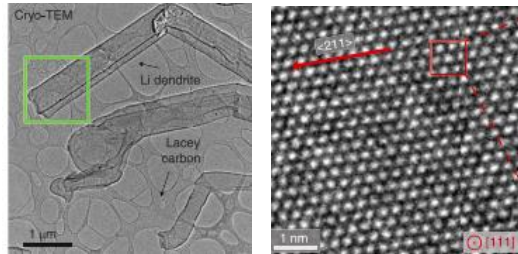


Chaojiang Niu Jun Liu



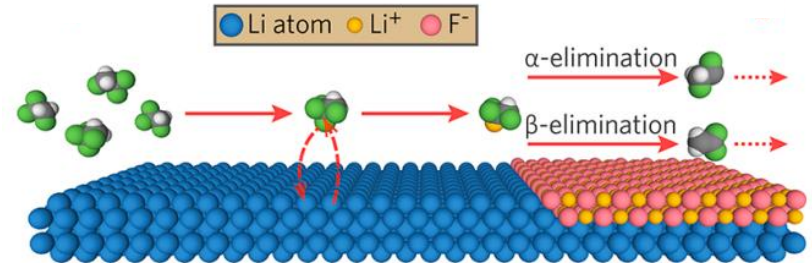
# Technical Accomplishments

*Reveal the atomic structure of Li metal by cryo-electron microscopy*



*Science*, 358, 506 (2017)

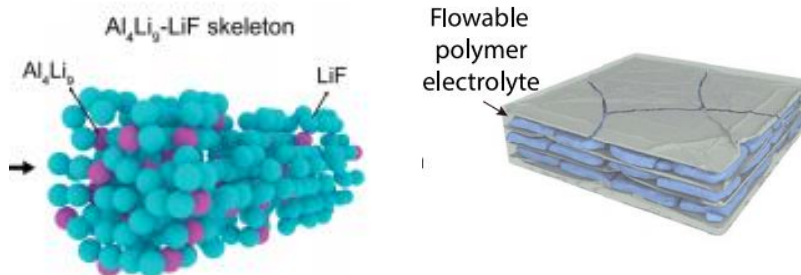
*Form stable artificial SEI layer via gas-phase reaction*



*J. Am. Chem. Soc.*, **139**, 11550 (2017)

*Nano Lett.* **17**, 3731 (2017); *Nano Lett.* **17**, 5171 (2017); *ACS Nano* **11**, 7019 (2017)

*Form stable host for Li metal: Li-SiO composite*

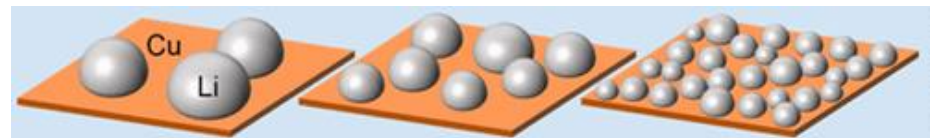


*Proc. Natl. Acad. Sci.*, **114**, 4613 (2017)

*Sci. Adv.* **3**, e1701301 (2017)

*Sci. Adv.* **3**, eaao0713 (2017)

*Selective deposition and stable encapsulation of lithium through heterogeneous seeded growth*



*Nat. Energy* **1**, 16010 (2016)

*Proc. Natl. Acad. Sci.*, **114**, 12138 (2017)

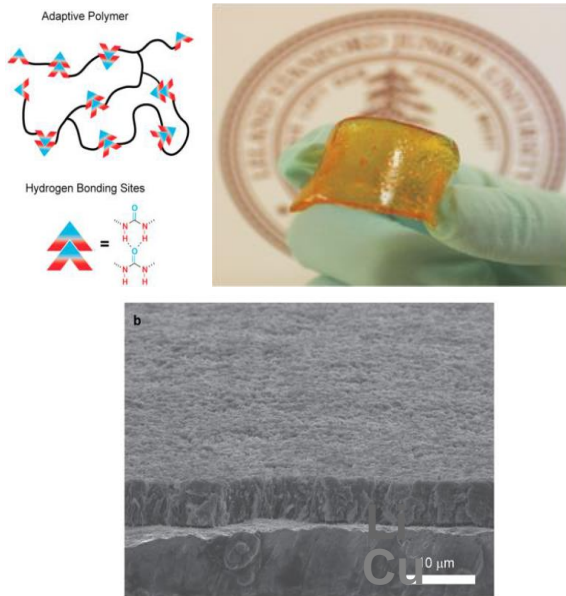
*Nano Lett.* **17**, 1132 (2017)

# Technical Accomplishments

## Soft, dynamic polymer coatings for Li

### Self-healing polymer (SHP)

- Low glass transition temperature
- Promotes uniform Li deposition

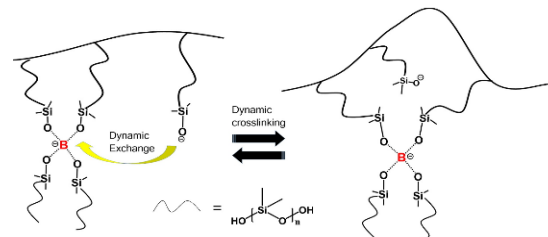


*A highly viscoelastic polymer coated on Li metal anode leads to uniform Li deposition*

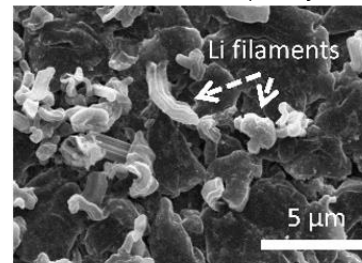
Bao, Z., Cui, Y., et al. *ACS Energy Lett.* 1247–1255 (2016)

### Silly Putty (SP)

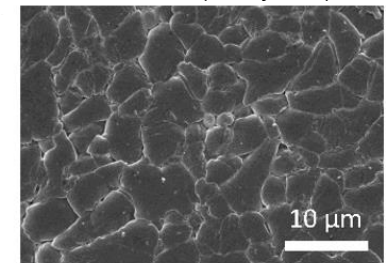
- Liquid-like at rest
- Solid-like upon stress



Without SP (75 cycles)



With SP (75 cycles)

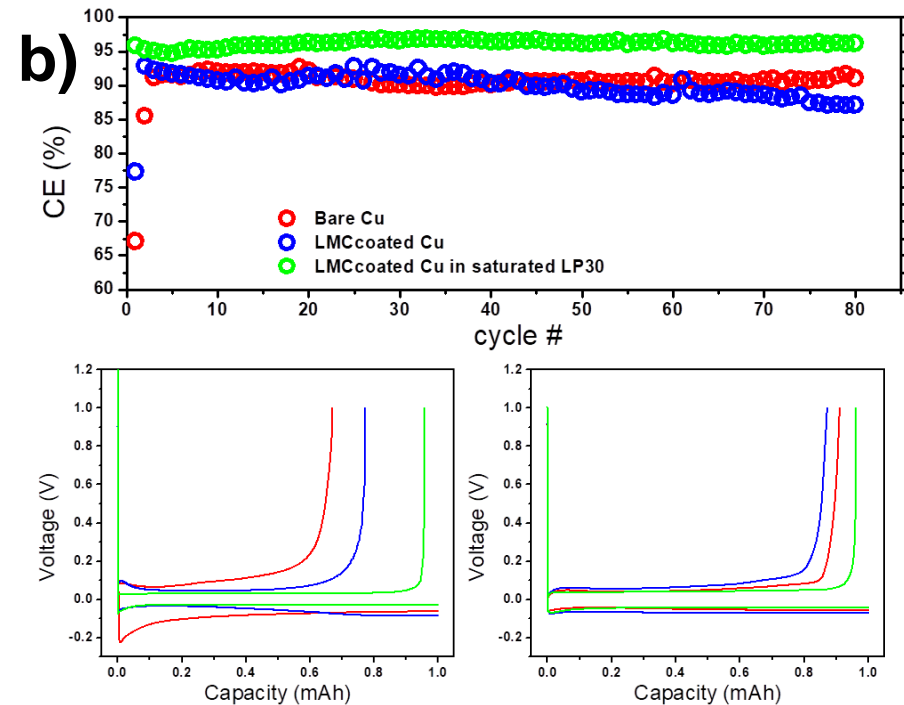
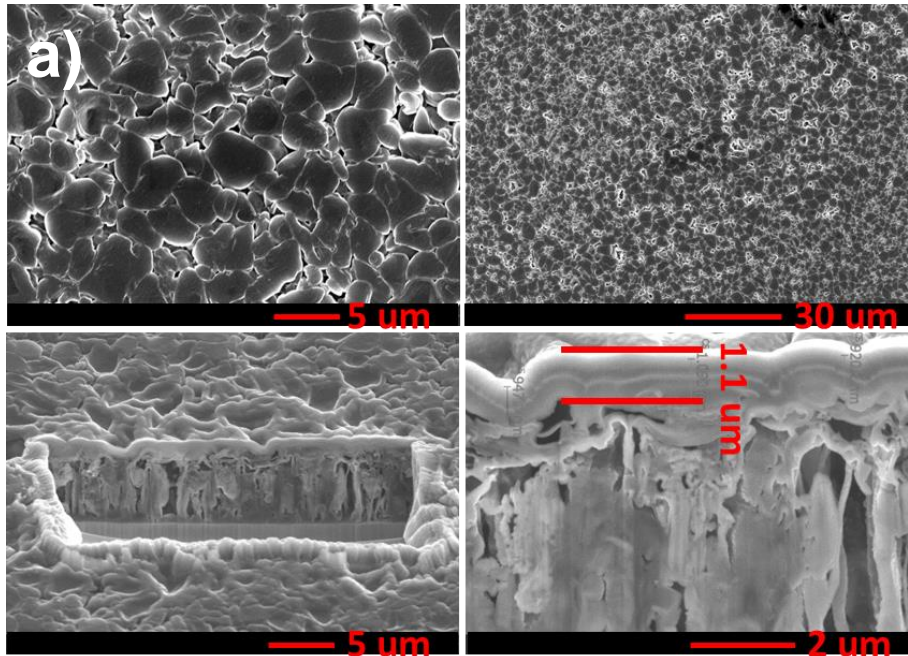


*Cross-linked polymer enables adaptive interfacial layer for Li metal anodes.*

Bao, Z., Cui, Y., et al. *J. Am. Chem. Soc.* 139 4815 (2017)

# Technical Accomplishments

## Li-methyl Carbonate Protection Layer



- FIB-SEM shows the LMC coating layer is around 1.1 μm that covers the plated Li
- The CE of bare Cu in LP30 is 90.5%, while the CE of LMC coated Cu in LMC saturated LP30 is improved to 96.1%. LP30: 1M LiPF<sub>6</sub> in EC/DMC (1:1)



# Collaboration and Coordination with Other Institutions

## Partners:

- Stanford University: Coating and host for Li metal anode, cryo-electron microscopy
- SLAC In situ X-ray characterization of Li metal anode
- University of California at San Diego: protective coating and in situ cryo-TEM characterization
- Idaho National Laboratory: pressure effect on Li cycling and test of PNNL-made pouch cell
- University of Washington: Simulation of Li metal batteries
- UT Texas: Solid state electrolyte
- Army Research Lab: electrolyte development
- University of Maryland: electrolyte/interphase

# Remaining Challenges/Barriers

To enable a Li metal batteries with more than 500 Wh/kg, new electrolytes and protection layer will be developed to enable Li metal anode to operate at the following conditions at the same time:

- CE of Li metal  $> 99.9\%$
- Lean electrolyte condition ( $< 3\text{g/Ah}$ )
- Thin Li metal ( $\text{N/P} < 2$ )
- High capacity ( $> 4 \text{ mAh/cm}^2$ ) and high current ( $\text{C}/3$ )

# Proposed Future Work

- An fundamental understanding on the stability of Li metal anode in non-aqueous electrolyte will be developed through simulation and characterization.
- New formulation of LHCE will be developed to reduce the cost of electrolyte and improve their safety.
- Electrolyte additive or mixed solvent/mixed salts will de developed to satisfy the multiple requirement (such as high current density and lean electrolyte condition) at the same time.
- 3D conductive host will be combined with the optimized electrolytes to minimize the volume expansion of Li metal anode.
- The voltage stability of protective coating on Li metal anode will be tailored for Li/NMC cells.
- Better protection layer and/or separators will be developed to prevent Li dendrite growth in extreme conditions.

# Summary

- LHCE enables high efficiency cycling of both Li metal anode (up to 99.5%) and stable cycling of Li/NMC cells (>95% capacity retention after 300 cycles and 80% capacity retention after 700 cycles in coin cells).
- Non-flammable LHCE enabled  $\geq 300$  Wh/kg Li/NMC pouch cells with more than 100 stable cycles.
- Several effective protection layers, including self-healing polymer, gas induced LiF layer, Li-methyl carbonate Layer etc have been developed to enable smooth deposition of Li metal anode.
- Several stable lithium-hosts have been developed to minimize the volume expansion of Li based anode.
- Advanced characterization techniques, including cryo-electron microscopy, in situ X-Ray Diffraction and Scattering and X-Ray Absorption Spectroscopy have been developed to understand the performance of Li metal anode with unprecedented accuracy and in situ capability.

# Acknowledgments

- ✓ DOE/VTO/Battery500 program
- ✓ PNNL Team Members: Jie Xiao, Xiaodi Ran, Lu Yu, Chaojiang Niu, Shuru Chen; Xia Cao
- ✓ Battery 500 PIs and team members

